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(d) 4. Answer. Answer: (d) 4 Explanation: Each water molecule can form a maximum of four hydrogen bonds with neighboring water molecules. The two hydrogens of the water molecule can form hydrogen bonds with other oxygens in ice, and the two lone pair of electrons on oxygen of the water molecule can attract other hydrogens in ice.

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(d) 4 Explanation: Each
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a maximum of four
hydrogen bonds with
neighboring water
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hydrogens of the water
molecule can form
hydrogen bonds with
other oxygens in ice, and
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